
Imf Liquids And Solids Worksheet Answers

liquids / solids / imfs intermolecular forces (imfs) - liquids / solids / imfs intermolecular forces (imfs) attractions between molecules • not within molecules • not true bonds... weaker attractions • represented by dashed lines • physical properties (melting points, boiling points, vapor pressures, etc.) can be attributed to the amount and strength of imfs example: higher bp = stronger imfs **chapter10 imfs,solids & liquids - max study** - • liquids with large imf tend to be highly viscous. o modeling a liquid is difficult. gases have very small imfs and lots of motion. solids have very high imfs and next to no motion. liquids have both strong imfs and quite a bit of motion. types of solids • crystalline solids: highly regular arrangement of their components [often **solids, liquids, and intermolecular forces (imfs)** - solids, liquids, and intermolecular forces (imfs) this part of the unit will describe various attributes of solid and liquid state properties. it will use the microscale ideas of particle composition, shape, arrangement, and intermolecular forces to help explain the observed macroscopic properties of substances. the **ap* chemistry intermolecular forces liquids & solids - google** - intermolecular forces, liquids & solids 2 bonded h h "bond" o hydrogen bonding—the force of attraction between the hydrogen atom of one molecule and an unshared electron pair on f, o, or n of a neighboring molecule (a special case of dipole-dipole). this is the strongest imf. never confuse hydrogen bonding with a bonded hydrogen. the **chapter 11 intermolecular forces, liquids, and solids** - intermolecular forces explaining macroscopic behavior: viscosity • viscosity is the resistance to flow in liquids • viscosity is dependent on more than just molecule size - it also depends on the kinds of attractive forces between molecules • viscosities of various liquids at 20°C (in centipoise) liquid viscosity (cp) type of im attraction **ap ws imfs solids key - conejo valley unified school district** - whs ap chemistry imf's, liquids, & solids imf's in solids indicate the strongest imf holdin to ether c stals of the followin ionic crystal network solid

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